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## Structure Reports

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## Yang-Gen Hu,* Ai-Hua Zheng and Gui-Hua Li

Department of Medicinal Chemistry, Yunyang Medical College, Shiyan 442000, People's Republic of China

Correspondence e-mail:
huyangg111@yahoo.com.cn

## Key indicators

Single-crystal X-ray study
$T=292 \mathrm{~K}$
Mean $\sigma(\mathrm{C}-\mathrm{C})=0.002 \AA$
$R$ factor $=0.043$
$w R$ factor $=0.117$
Data-to-parameter ratio $=16.2$
For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

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## 3-(4-Methylphenyl)-2-(1-pyridyl)-3H-benzo-[4,5]furo[3,2-d]pyrimidin-4-one

In the crystal structure of the title compound, $\mathrm{C}_{22} \mathrm{H}_{21} \mathrm{~N}_{3} \mathrm{O}_{2}$, the packing of the molecules is mainly governed by intermolecular $\mathrm{C}-\mathrm{H} \cdots \mathrm{O}$ hydrogen bonds and $\pi-\pi$ interactions.

## Comment

In recent years, we have been engaged in the preparation of heterocyclic derivatives containing a fused pyrimidinone system using the aza-Wittig reaction (Ding, Chen \& Huang, 2004; Ding, Xu \& Zhao, 2004; Ding et al. 2005). Some X-ray crystal structure reports for benzofuro[3,2- $d$ ]pyrimidine derivatives have been published (Hu, Li et al., 2005; $\mathrm{Hu}, \mathrm{Xu}$ et al., 2005; Sun \& Hu, 2005). We describe here the structure of the title compound, (I), prepared using an iminophosphorane with 4-methylphenyl isocyanate and by a subsequent reaction with piperidine under mild conditions.

(I)

(II)

The three fused rings of the benzofuro[3,2- $d$ ]pyrimidine system are essentially coplanar (Fig. 1). The piperidine ring shows a distorted chair conformation $\left[\varphi=203.84(2)^{\circ}\right.$ and $\theta=$ $175.54(2)^{\circ}$, puckering amplitude $=0.583(2) \AA$; Cremer \& Pople, 1975]. An intermolecular C-H. . O hydrogen bond between the ketone group and the methyl-substituted benzene ring (Table 2 and Fig. 2) is effective in stabilizing the crystal structure. There are also intermolecular $\pi-\pi$ interactions (Janiak, 2000). In the benzofuro[3,2-d]pyrimidine system, the maximum and minimum interplanar perpendicular distances between adjacent rings related by an inversion centre are 3.400 (3) and 3.650 (3) $\AA$, respectively, the centroid-tocentroid distances are 3.613 (3) and 3.461 (3) $\AA$, and the dihedral angles are 0.00 (2) and 2.57 (2) ${ }^{\circ}$ (Fig. 2).

## Experimental

To a solution of iminophosphorane ( $1.40 \mathrm{~g}, 3 \mathrm{mmol}$ ) in dry dichloromethane ( 15 ml ) was added 4-methylphenyl isocyanate ( 3 mmol ) under nitrogen at room temperature. After reaction, the mixture was allowed to stand for 10 h at 273-278 K, the solvent was removed under reduced pressure and diethyl ether/petroleum ether (1:2 $\mathrm{v} / \mathrm{v}, 20 \mathrm{ml})$ was added to precipitate triphenylphosphine oxide.

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After filtration, the solvent was removed to give the carbodiimide 3-p-tolyliminomethyleneaminobenzofuran-2-carboxylic acid ethyl ester, (II), which was used directly without further purification. To a solution of (II) in dichloromethane ( 15 ml ) was added piperidine $(3 \mathrm{mmol})$. After the reaction mixture was allowed to stand for 0.5 h , the solvent was removed and anhydrous ethanol ( 10 ml ) with several drops of EtONa in EtOH was added. The mixture was stirred for 2 h at room temperature. The solution was concentrated under reduced pressure and the residue was recrystallized from ethanol to give the title compound, (I). The product was recrystallized from ethanol-dichloromethane ( $1: 1 \mathrm{v} / \mathrm{v}, 20 \mathrm{ml}$ ) at room temperature to give crystals suitable for X-ray diffraction (m.p. 485 K , yield $82 \%$ ).

## Crystal data

$\mathrm{C}_{22} \mathrm{H}_{21} \mathrm{~N}_{3} \mathrm{O}_{2}$
$M_{r}=359.42$
Monoclinic, $P 2_{1} / n$
$a=14.4774$ (12) A
$b=7.7044$ (7) A
$c=16.6859$ (15) A
$\beta=97.754$ (2) ${ }^{\circ}$
$V=1844.1$ (3) $\AA^{3}$
$Z=4$

## Data collection

Bruker SMART 4K CCD area-
detector diffractometer
$\varphi$ and $\omega$ scans
Absorption correction: none
10894 measured reflections
3995 independent reflections
$D_{x}=1.295 \mathrm{Mg} \mathrm{m}^{-3}$
Mo $K \alpha$ radiation
Cell parameters from 3667 reflections
$\theta=2.5-27.3^{\circ}$
$\mu=0.09 \mathrm{~mm}^{-1}$
$T=292$ (2) K
Block, colourless
$0.30 \times 0.30 \times 0.30 \mathrm{~mm}$

2960 reflections with $I>2 \sigma(I)$
$R_{\text {int }}=0.052$
$\theta_{\text {max }}=27.0^{\circ}$
$h=-15 \rightarrow 18$
$k=-9 \rightarrow 9$
$l=-21 \rightarrow 17$

## Refinement

Refinement on $F^{2}$
$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.043$
$w R\left(F^{2}\right)=0.117$
$S=0.98$
3995 reflections
246 parameters
H -atom parameters constrained
$w=1 /\left[\sigma^{2}\left(F_{o}^{2}\right)+(0.0668 P)^{2}\right]$
where $P=\left(F_{\mathrm{o}}^{2}+2 F_{\mathrm{c}}^{2}\right) / 3$
$(\Delta / \sigma)_{\text {max }}<0.001$
$\Delta \rho_{\text {max }}=0.16$ e $\AA^{-3}$
$\Delta \rho_{\min }=-0.22$ e $\AA^{-3}$
Extinction correction: SHELXL97 (Sheldrick, 1997)
Extinction coefficient: 0.023 (2)

Table 1
Selected geometric parameters $\left(\AA^{\circ},{ }^{\circ}\right)$.

| C1-N1 | $1.4735(18)$ | C14-N2 | $1.4243(16)$ |
| :--- | ---: | :--- | ---: |
| C5-N1 | $1.4715(16)$ | C15-O2 | $1.3791(15)$ |
| C6-N3 | $1.3015(15)$ | C16-N3 | $1.3694(16)$ |
| C6-N1 | $1.3904(16)$ | C18-O2 | $1.3852(15)$ |
| C6-N2 | $1.3961(15)$ | C7-N2 | $1.4501(15)$ |
| C14-O1 | $1.2174(15)$ |  |  |
| N1-C1-C2 | $110.02(12)$ | C15-C16-N3 | $123.87(11)$ |
| N1-C5-C4 | $110.02(11)$ | N3-C16-C17 | $129.97(12)$ |
| N3-C6-N1 | $120.37(10)$ | C19-C18-O2 | $125.01(13)$ |
| N3-C6-N2 | $123.83(11)$ | O2-C18-C17 | $111.55(11)$ |
| N1-C6-N2 | $115.76(10)$ | C8-C7-N2 | $120.29(11)$ |
| O1-C14-C15 | $128.27(12)$ | C6-N3-C16 | $115.31(10)$ |
| O1-C14-N2 | $121.65(11)$ | C6-N2-C14 | $123.33(10)$ |
| C15-C14-N2 | $110.04(10)$ | C6-N1-C5 | $115.10(10)$ |
| C16-C15-O2 | $112.87(11)$ | C15-O2-C18 | $104.38(10)$ |
|  |  |  |  |
| N2-C14-C15-C16 | $-3.14(18)$ | N1-C6-N3-C16 | $-179.70(11)$ |
| N2-C14-C15-O2 | $177.82(11)$ | N3-C6-N2-C14 | $2.12(19)$ |
| O2-C15-C16-N3 | $-177.53(11)$ | N1-C6-N2-C7 | $6.34(17)$ |
| C20-C19-C18-O2 | $179.72(12)$ | N3-C6-N1-C5 | $18.96(17)$ |
| C22-C17-C18-O2 | $-179.73(11)$ | C14-C15-O2-C18 | $179.87(12)$ |
| C16-C17-C18-O2 | $0.96(14)$ | C17-C18-O2-C15 | $-1.05(13)$ |
| N2-C7-C8-C9 | $-178.73(12)$ |  |  |



Figure 1
View of the molecular structure of (I), showing the atom-labelling scheme. Displacement ellipsoids are drawn at the $50 \%$ probability level.


Figure 2
Packing view of (I), showing the $\mathrm{C}-\mathrm{H} \cdots \mathrm{O}$ hydrogen bonds (dashed lines) and the $\pi-\pi$ stacking interactions. H atoms attached to C atoms have been omitted for clarity. The suffixes $\mathrm{a}, \mathrm{b}$ and c correspond to symmetry $\operatorname{codes}\left(\frac{3}{2}-x, \frac{1}{2}+y, \frac{1}{2}-z\right),\left(\frac{3}{2}-x, \frac{1}{2}-y, \frac{1}{2}-z\right)$ and $(2-x, 1-y$, $1-z$ ), respectively.

Table 2
Hydrogen-bond geometry ( $\left(\AA,{ }^{\circ}\right)$.

| $D-\mathrm{H} \cdots A$ | $D-\mathrm{H}$ | $\mathrm{H} \cdots A$ | $D \cdots A$ | $D-\mathrm{H} \cdots A$ |
| :--- | :--- | :--- | :--- | :--- |
| $\mathrm{C} 8-\mathrm{H} 8 \cdots \mathrm{O}_{1}{ }^{\mathrm{i}}$ | 0.93 | 2.51 | $3.2369(16)$ | 135 |

Symmetry code: (i) $-x+\frac{3}{2}, y+\frac{1}{2},-z+\frac{1}{2}$.
The H atoms were positioned geometrically $[\mathrm{C}-\mathrm{H}=0.93(\mathrm{CH})$, $0.97\left(\mathrm{CH}_{2}\right)$ and $\left.0.96 \AA\left(\mathrm{CH}_{3}\right)\right]$ and constrained to ride on their parent atoms, with $U_{\text {iso }}(\mathrm{H})=1.2\left(1.5\right.$ for methyl) times $U_{\text {eq }}(\mathrm{C})$.

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Data collection: SMART (Bruker, 2001); cell refinement: SAINT (Bruker, 2001); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL (Sheldrick, 2001); software used to prepare material for publication: SHELXTL.

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## References

Bruker (2001). SMART and SAINT. Bruker AXS Inc., Madison, Wisconsin, USA.

Cremer, D. \& Pople, J. A. (1975). J. Am. Chem. Soc. 97, 1354-1358.
Ding, M.-W., Chen, Y.-F. \& Huang, N.-Y. (2004). Eur. J. Org. Chem. 18, 38723878.

Ding, M.-W., Huang, N.-Y. \& He, H.-W. (2005). Synthesis, 10, 16011604.

Ding, M.-W., Xu, S.-Z. \& Zhao, J.-F. (2004). J. Org. Chem. 69, 83668371.

Hu, Y.-G., Li, G.-H., Tian, J.-H., Ding, M.-W. \& He, H.-W. (2005). Acta Cryst. E61, o3266-o3268.
Hu, Y.-G., Xu, S.-Z., Yuan, J.-Z., Ding, M.-W. \& He, H.-W. (2005). Acta Cryst. E61, o2649-o2651.
Janiak, C. (2000). J. Chem. Soc. Dalton Trans. pp. 3885-3896.
Sheldrick, G. M. (1997). SHELXL97 and SHELXS97. University of Göttingen, Germany.
Sheldrick, G. M. (2001). SHELXTL. Version 5.0. Bruker AXS Inc., Madison, Wisconsin, USA.
Sun, Y. \& Hu, Y.-G. (2005). Acta Cryst. E61, o4233-o4235.


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